Naming and Writing Acids

All acids begin with "H"

* Write formulas using criss-cross rule! *

"H" + "anion" = acid

$$H^{+'} + F^{-} = HF$$

$$H^{+1} + (50^{-3}) = H_2 SO_4$$

Feb 8-9:39 AM

Binary Acids: (H and some single element from P.T.)

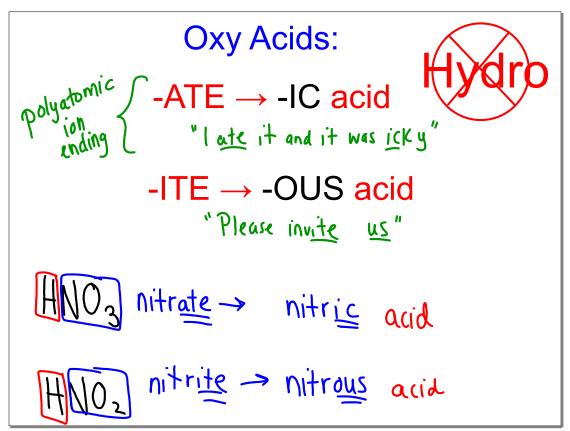
- 1) must begin name w/ HYDRO
- 2) add the suffix -ic acid to the anion



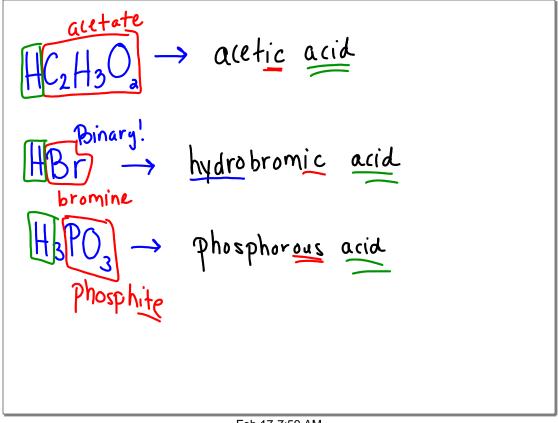
hydrofluoric acid







Feb 8-9:59 AM



Feb 17-7:59 AM

Acids Notes

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hydroselenic acid H+1 Se<sup>-2</sup> → H<sub>2</sub>Se

Look on
the P.T.!

Binary!

Oxalic acid H+1 C<sub>2</sub>O<sub>4</sub> → H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>

Oxalate

No Hydro = Polyatomic!
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Feb 17-8:55 AM