STOICHIOMETRY: VOLUME-VOLUME PROBLEMS

Name _____

1. $N_2 + 3H_2 \rightarrow 2NH_3$

What volume of hydrogen is necessary to react with five liters of nitrogen

2. What volume of ammonia is produced in the reaction in Problem 1? (Use 5 L of N_2)

3. $C_3H_8 + 5O_2 \rightarrow 3CO_2 + 4H_2O$

If 20 liters of oxygen are consumed in the above reaction, how many liters of carbon dioxide are produced?

4. $2H_2O \rightarrow 2H_2 + O_2$

If 30 L of hydrogen are produced in the above reaction, how many iliters of oxygen are produced?

5. $2CO + O_2 \rightarrow 2CO_2$

How many liters of carbon dioxide are produced if 75 liters of carbon monoxide are burned ? How many liters of oxygen are necessary?